



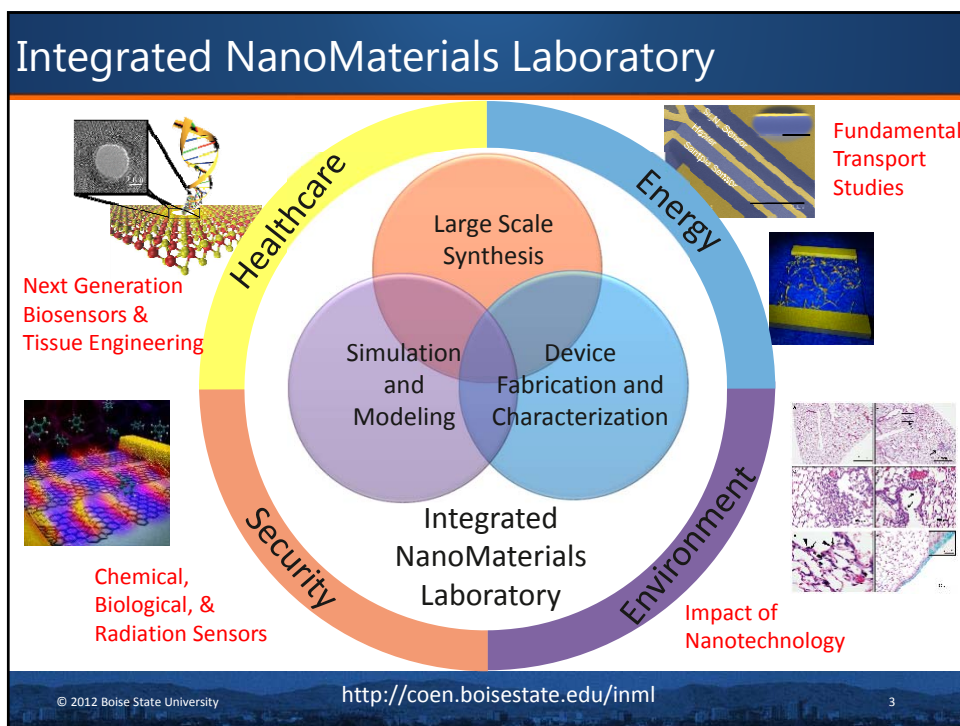
# Part 1: A brief Introduction to Graphene Properties, Synthesis, and Applications

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Assistant Professor  
Materials Science and Engineering

## Global Energy Outlook



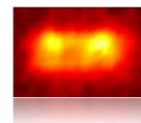
(adapted from A. Majumdar, U.S. and Japan Seminar on Nanoscale Heat Transfer, 2007)



## Outline of the talk

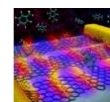
### Part 1: A brief Introduction to Graphene Properties, Synthesis, and Applications

- Review of bonding and structure
- Energy Bands in Solids
- Microscopy Techniques
- Graphene Synthesis and Applications



### Part 2: Atomic Layer Research in the INML

- Carbon nanotube transistors
- Role of defects in graphene sensing and transport
- Biomolecules to Tissue Engineering in the INML



## Atomic Structure

- electron  $9.11 \times 10^{-31} \text{ kg}$
- proton  $1.67 \times 10^{-27} \text{ kg}$
- neutron  $1.67 \times 10^{-27} \text{ kg}$
- Atomic # = number of protons in the nucleus or atom  
= of electrons in a neutral species
- Isotope: Determined by number of neutrons in atom
- Ion: Charged atom, unequal number of electrons and protons
- amu = 1/12 mass of  $^{12}\text{C}$  isotope
- Atomic wt = wt of  $6.023 \times 10^{23}$  molecules or atoms,  
weighted average of all isotopes
- 1 amu/atom = 1 g/mole

Atomic Number

14

Symbol

Si

Atomic Weight

28.08

$$\bar{A}_M = \sum_i f_{iM} A_{iM}$$

## Valence Electrons

- Valence Electrons determine all of the following properties
  - Chemical
  - Thermal
  - Optical
  - Electrical
- Valence electrons – those in unfilled shells
  - Filled shells are more stable
- Valence electrons are most available for bonding
- Example: Carbon (atomic number = 6)
  - $1s^2 2s^2 2p^2$

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- Most elements: Electron configuration **not stable**.

Element	Atomic #	Electron configuration
Hydrogen	1	$1s^1$
Helium	2	$1s^2$ (stable)
Lithium	3	$1s^2 2s^1$
Beryllium	4	$1s^2 2s^2$
Boron	5	$1s^2 2s^2 2p^1$
Carbon	6	$1s^2 2s^2 2p^2$
...	...	...
Neon	10	$1s^2 2s^2 2p^6$ (stable)
Sodium	11	$1s^2 2s^2 2p^6 3s^1$
Magnesium	12	$1s^2 2s^2 2p^6 3s^2$
Aluminum	13	$1s^2 2s^2 2p^6 3s^2 3p^1$
...	...	...
Argon	18	$1s^2 2s^2 2p^6 3s^2 3p^6$ (stable)
...	...	...
Krypton	36	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6$ (stable)

Adapted from Table 2.2,  
Callister & Rethwisch 9e.

- Why? **Valence** (outer) shell usually not filled completely.

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## Bonding and the Periodic Table

give up 1e

give up 2e

give up 3e

accept 2e

accept 1e

inert gases

IA		IIA		VIII										IB	IIB	IIIA	IVA	VA	VIA	VIIA	0																																																																																																																																																																																																										
1	H	2	He	3	B	4	C	5	N	6	O	7	F	8	Ne	9	Na	10	Mg	11	Al	12	Si	13	P	14	S	15	Cl	16	Ar	17	K	18	Ca	19	Sc	20	Ti	21	V	22	Cr	23	Mn	24	Fe	25	Co	26	Ni	27	Cu	28	Zn	29	Ga	30	Ge	31	As	32	Se	33	Br	34	Kr	35	Rb	36	Sr	37	Y	38	Zr	39	Nb	40	Mo	41	Tc	42	Ru	43	Rh	44	Pd	45	Ag	46	Cd	47	In	48	Sn	49	Sb	50	Te	51	I	52	Xe	53	Cs	54	Ba	55	Ra	56	Ac	57	Rf	58	Db	59	Sg	60	Bh	61	Hs	62	Mt	63	Ds	64	Rf	65	Db	66	Sg	67	Bh	68	Hs	69	Mt	70	Ds	71	Rf	72	Db	73	Sg	74	Bh	75	Hs	76	Mt	77	Ds	78	Rf	79	Db	80	Sg	81	Bh	82	Hs	83	Mt	84	Ds	85	Rf	86	Db	87	Sg	88	Bh	89	Hs	90	Mt	91	Ds	92	Rf	93	Db	94	Sg	95	Bh	96	Hs	97	Mt	98	Ds	99	Rf	100	Db	101	Sg	102	Bh	103	Hs	104	Mt	105	Ds	106	Rf	107	Db	108	Sg	109	Bh	110	Hs	111	Mt	112	Ds

←

Electropositive elements: Readily give up electrons to become + ions.

→

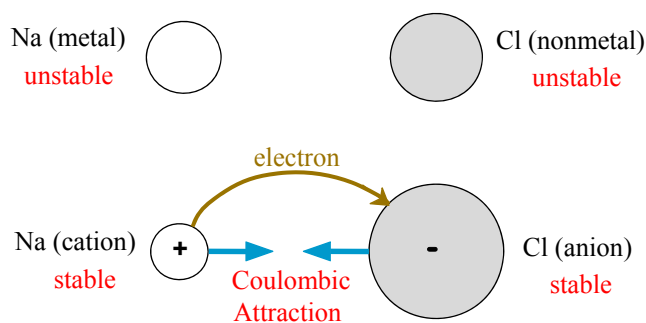
Electronegative elements: Readily acquire electrons to become - ions.

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## Ionic Bonding

- Large difference in electronegativity is required
- Occurs between **+ ions and – ions**
- Requires **electron transfer**



## Covalent Bonding

- Sharing of electrons results in strong bonds
- Similar **electronegativity**  $\therefore$  **share electrons**
- Bonds determined by valence

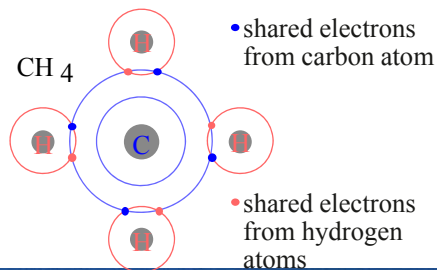
– **s & p orbitals dominate bonding**

C: has 4 valence  $e^-$ ,  
needs 4 more

H: has 1 valence  $e^-$ ,  
needs 1 more

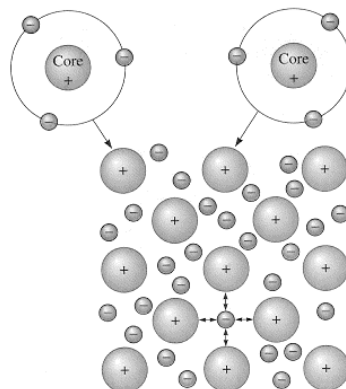
- **Methane: CH<sub>4</sub>**

**Electronegativities  
are comparable.**



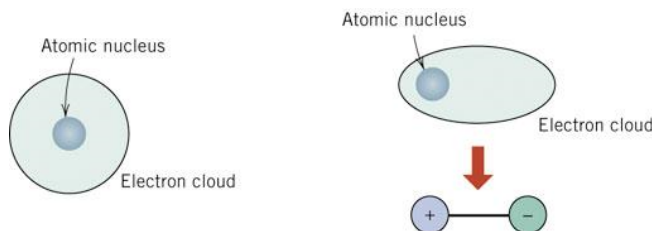
## Metallic Bonding

- 80 of the ~109 natural elements are metals
- Valence electrons are released from the atom leaving behind a **positively charged ion core**, called a “cation”
- Cations are held together by a **sea of electrons**
- the bond occurs because the positive cations are attracted to the negative electrons.

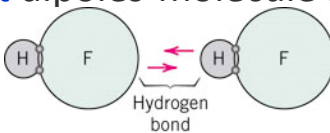


## Secondary bonds

- **Fluctuating** dipoles

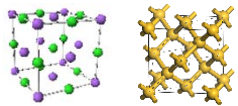
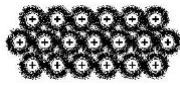
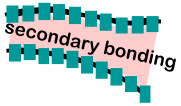


- **Permanent** dipoles-molecule induced




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Type	Bond Energy	Comments
Ionic	Large!	Nondirectional (ceramics)
Covalent	Variable <i>large-Diamond</i> <i>small-Bismuth</i>	Directional (semiconductors, ceramics polymer chains)
Metallic	Variable <i>large-Tungsten</i> <i>small-Mercury</i>	Nondirectional (metals)
Secondary	smallest	Directional inter-chain (polymer) inter-molecular

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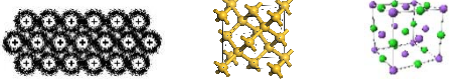
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<p><b>Ceramics and Semiconductors</b> (Ionic &amp; covalent bonding):</p> 	<p><b>Large bond energy</b> large <math>T_m</math> large <math>E</math> small <math>\alpha</math></p>
<p><b>Metals</b> (Metallic bonding):</p> 	<p><b>Variable bond energy</b> moderate <math>T_m</math> moderate <math>E</math> moderate <math>\alpha</math></p>
<p><b>Polymers</b> (Covalent &amp; Secondary):</p> 	<p><b>Directional Properties</b> Secondary bonding dominates small <math>T_m</math> small <math>E</math> large <math>\alpha</math></p>

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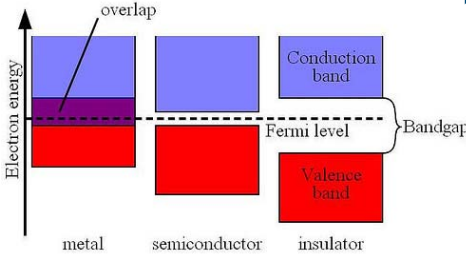
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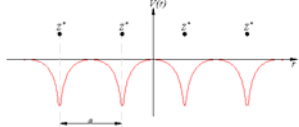
## Bonding and Crystal Structure



$$\left[ -\frac{\hbar^2}{2m} \nabla^2 + U(\mathbf{r}) \right] \psi(\mathbf{r}) = E\psi(\mathbf{r})$$

Electrical properties span 20 orders of magnitude! ➔






$$\left( \frac{\hbar^2}{2m} k^2 - E \right) c_k + \sum_G U_G c_{k-G} = 0.$$

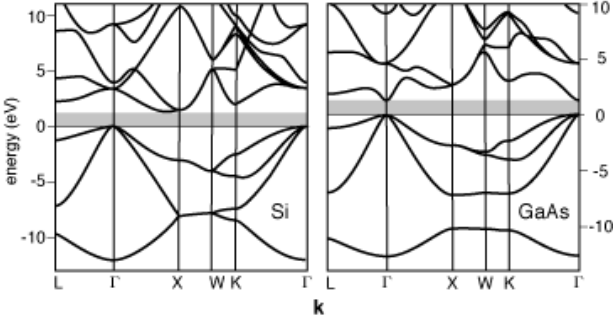
- Properties of a material depend on the crystal structure.
- Electrons in a crystal are subject to periodic potentials.

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## Energy Bands in Solids



$$\left( \frac{\hbar^2}{2m} k^2 - E \right) c_k + \sum_G U_G c_{k-G} = 0.$$

- Parabolic relation between energy and momentum
- Effective mass of an electron relates to the curvature of the band
- Forbidden energy states!

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## Microscopy

Optical resolution ~ diffraction limited

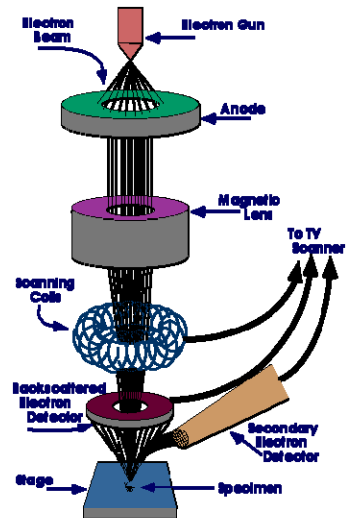
For higher resolution need higher frequency

- X-Rays? Difficult to focus.
- Electrons
  - wavelengths ca. 3 pm (0.003 nm)
    - (Magnification - 1,000,000X)
  - Atomic resolution possible
  - Electron beam focused by magnetic lenses.
- Atomic Forces
  - Sensing atomic interactions between a cantilever and a surface allows for direct correlation of structure and properties

## Scanning Electron Microscopy



Analysis of scattered electrons can produce images, crystallographic information, and elemental analysis



## Scanning Electron Images

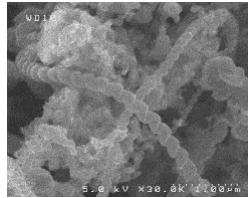
### Dislocations



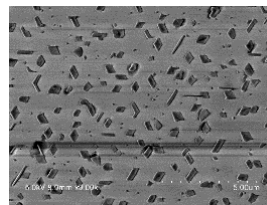
0.2 μm

Fig. 4.7, Callister & Rethwisch 9e.  
(Courtesy of M. R. Plichta, Michigan Technological University.)

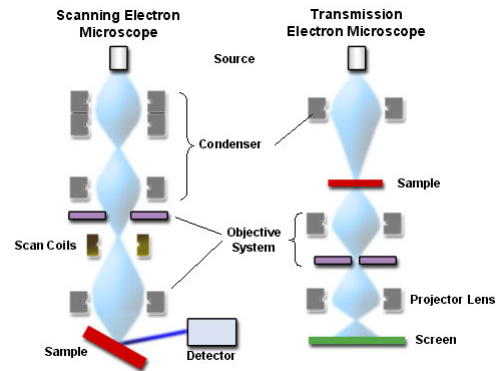
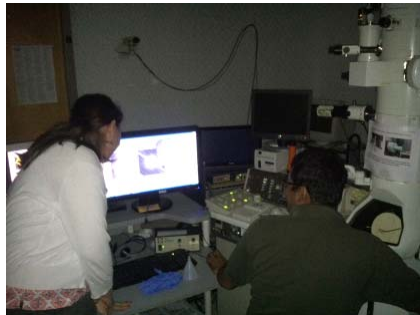
### Helical carbon nanotubes



### 2-D TMDC crystals on sapphire



## Transmission Electron Microscopy

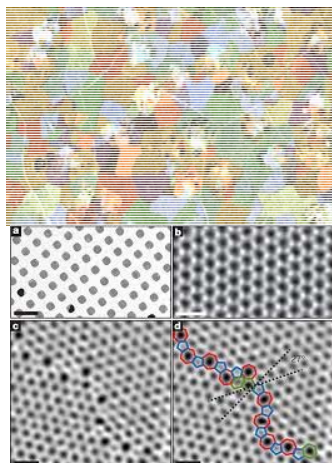


- TEM uses transmitted electrons to produce an image rather than secondary electrons
- Transmission through the sample produces diffraction patterns
- Higher energy → Higher Resolution

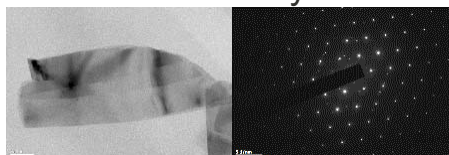


## Transmission Electron Microscope Images

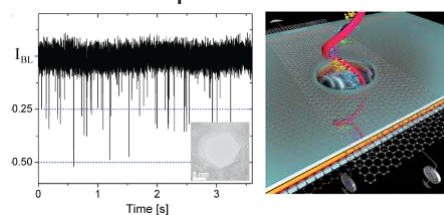
### Graphene Grain Boundaries



### 2-D TMDC crystals



### Nanopore Formation



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## Scanning Tunneling Microscopy

- Developed by Gerd Binnig and Heinrich Rohrer at the IBM Zurich Research Laboratory in 1982.

Binnig



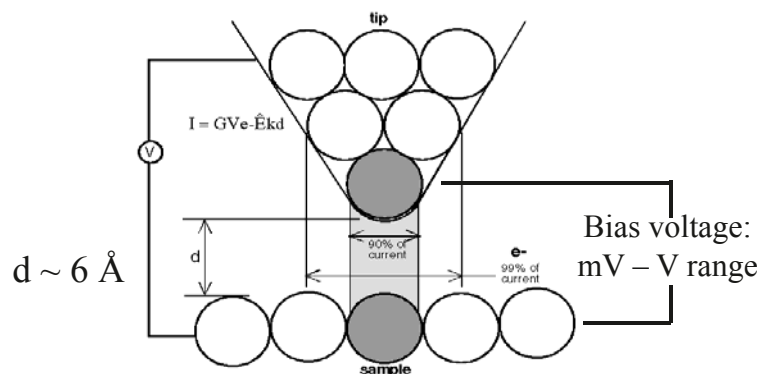
Rohrer

- The two shared half of the 1986 Nobel Prize in physics for developing STM.
- STM has fathered a host of new atomic probe techniques: Atomic Force Microscopy, Scanning Tunneling Spectroscopy, Magnetic Force Microscopy, Scanning Acoustic Microscopy, etc.

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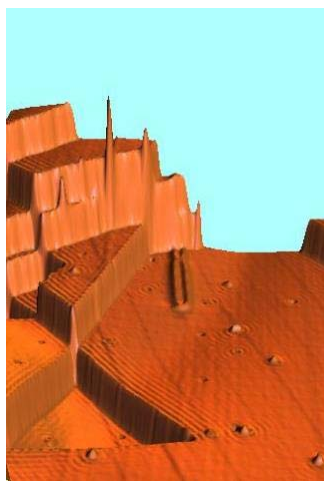
## Basic Principles of STM



Electrons tunnel between the tip and sample, a small current  $I$  is generated (10 pA to 1 nA).

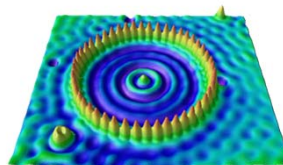
$I$  proportional to  $e^{-\kappa d}$ ,  $I$  decreases by a factor of 10 when  $d$  is increased by 1  $\text{\AA}$ .

## Interesting Images with STM



Copper Surface

Carbon monoxide molecules arranged on a platinum (111) surface.



Iron atoms on the surface of Cu(111)

## Atomic Force Microscopy

- Developed by Gerd Binnig, Calvin Quate, and Christoph Gerber in 1986



Binnig



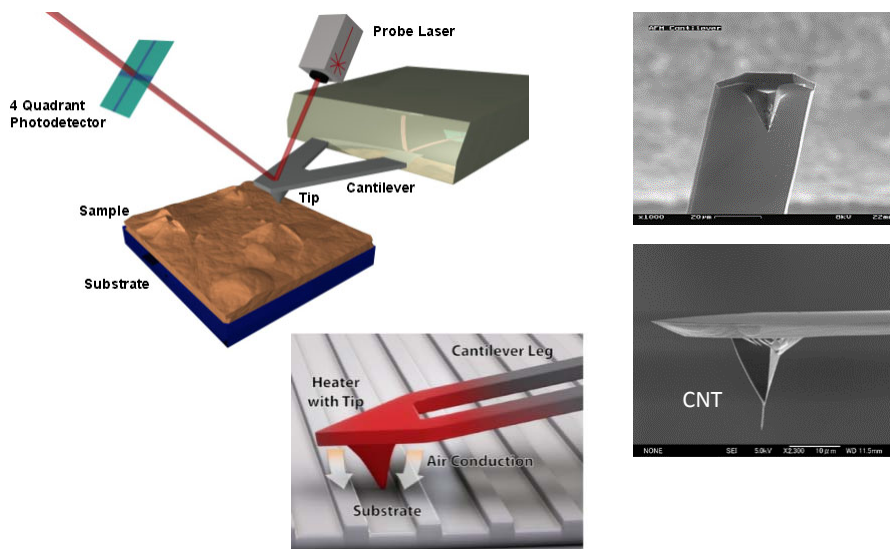
Quate



Gerber

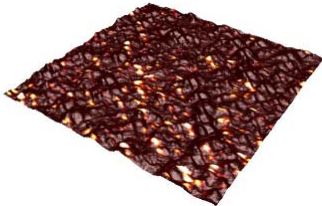
- First commercial AFM was introduced in 1989.
- AFM tips can be “functionalized” to probe a variety of physical properties with nanoscale resolution

## Atomic Force Microscopy

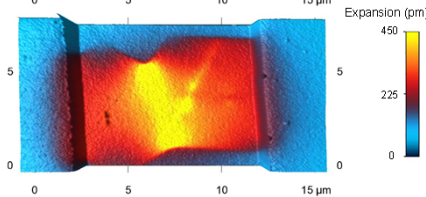


## Atomic Force Microscopy

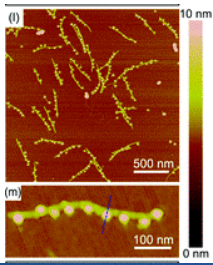
MoS<sub>2</sub> 500 nm x 500 nm scan



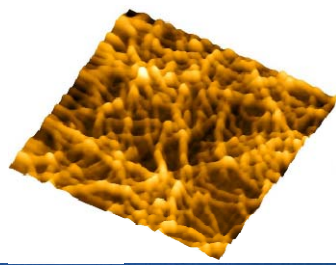
Scanning Joule Expansion Microscopy




DNA Origami with Au NPs (Bui, et al. Nano Lett. 2010)



RecA coated DNA (1 x 1 um scan)

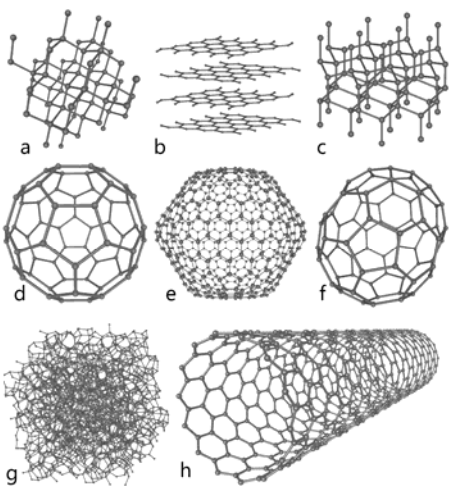


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## Carbon Allotropes



5 <b>B</b> <small>Boron</small> 10.81	6 <b>C</b> <small>Carbon</small> 12.01	7 <b>N</b> <small>Nitrogen</small> 14.01
13 <b>Al</b> <small>Aluminum</small> 26.98	14 <b>Si</b> <small>Silicon</small> 28.09	15 <b>P</b> <small>Phosphorus</small> 30.97
31 <b>Ga</b> <small>Gallium</small> 69.72	32 <b>Ge</b> <small>Germanium</small> 72.64	33 <b>As</b> <small>Arsenic</small> 74.92

A. Diamond  
 B. Graphite  
 C. lonsdaleite  
 D. C60 Buckminster Fullerene  
 E. C540  
 F. C70  
 G. Amorphous Carbon  
 H. Single-Wall Carbon Nanotube

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## Graphene: A 2D Building Block

Geim and Novoselov, Nature Materials, 2007

- If you can isolate a single sheet from graphite you get graphene.
- Physical properties return to bulk graphite after 10 layers.

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## Crystal Structure of Graphene

$$\left[ -\frac{\hbar^2}{2m} \nabla^2 + U(\mathbf{r}) \right] \psi(\mathbf{r}) = E\psi(\mathbf{r})$$

$$E(k_x, k_y) = \pm \gamma_0 \sqrt{1 + 4 \cos \frac{\sqrt{3} k_x a}{2} \cos \frac{k_y a}{2} + 4 \cos^2 \frac{k_y a}{2}}$$

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## Energy Bands in Graphene

- Energy bands are linear near the K and K' points of the reciprocal lattice.
- Massless Dirac Fermions with a Fermi velocity of  $c/300 \sim 1 \times 10^6$  m/s.
- Energy bands are symmetric  $\rightarrow$  equal electron and hole mobilities.
- Bands touch at the "Dirac Point"  $\rightarrow$  no band gap, zero density of states.
- Graphene is semi-metal or zero-band gap semiconductor.

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## Graphene Properties and Applications

Graphene has great electrical, thermal, mechanical, and optical properties!

- Symmetric energy band  $\rightarrow$  equal electron & hole mobility, 10-100x higher than Si
- Large optical phonon energy (0.18 eV vs. 0.06 eV in Si)
- Atomically thin  $\rightarrow$  High optical transparency and surface to volume ratio
- Strong  $\sigma$  bonds  $\rightarrow$  high thermal conductivity  $k \approx 20 \times k_{Si}$
- Emerging applications in layer-by-layer (LBL) assembled vdW solids

### Field-Effect Transistors

K. Novoselov, et al. Science (2004)

M.-H. Bae, et al. Nano Lett. (2012)

### Interconnects and Sensors

R.-H. Kim et al., Nano Lett. (2011)

Salehi-Khojin, et al., Adv. Mat. (2012)

### LBL Assembled vdW Solids

I. Britnell, et al., Science (2012)

G. Gao et al., Nano Lett. (2012)

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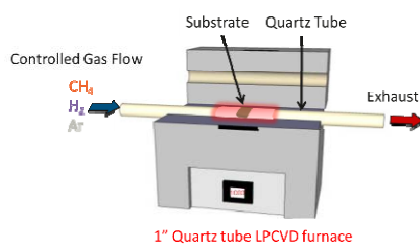


## Graphene Superlatives

- Thinnest imaginable material.
- Strongest material ever measured.
- Stiffest material known
- Most stretchable crystal.
- Record thermal conductivity.
- Highest current density at room temperature. ( $10^6 >$  copper)
- Highest intrinsic mobility. (100 times  $>$  Si)
- Lightest charge carriers. (zero rest mass)
- Longest mean free path.
- Most impermeable membrane.



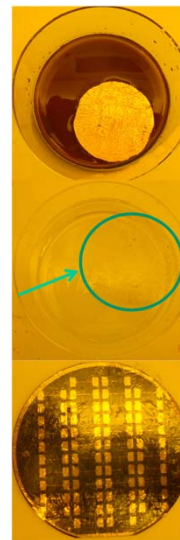
## CVD Growth



X.Li, et al, Science (2009)

- Graphene grown directly on Cu foil at 1000 °C
- CVD growth of graphene → polycrystalline films
- Graphene is transferred by etching of Cu foil in  $\text{FeCl}_3$
- Films are annealed in  $\text{Ar}/\text{H}_2$  at 400 °C to remove residue

A.Salehi-Khojin / D. Estrada, et al., *Advanced Materials* 24, 53-57 (2012)



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### Mechanical Exfoliation

Nobel 2010

Novoselov, K. S. et al., *Science* (2004)  
Novoselov, X. et al., *Science* (2009)

### Solvent Assisted Exfoliation

Y. Hernandez, et al., *Nat. Nano* (2008)  
D. Li, et al., *Nat. Nano* (2008)

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## Graphene Quality and Synthesis

Novoselov, *Nature* (2012)

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## Key Applications

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## Electrostatic Doping of Graphene

Positive electrostatic potential lowers electron energy (or effectively raises Fermi level).

Hole-doped                      Neutral                      Electron-doped

Increasing electrostatic potential

Local carrier density and type can be changed by applying an electrostatic potential. Use in graphene Field-Effect Transistors (FET).

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## Graphene FETs

- 3-terminal FET fabricated using 'tape' method [1] from natural flake graphite and e-beam lithography.

[1] Novoselov et al., *Science* (2004)

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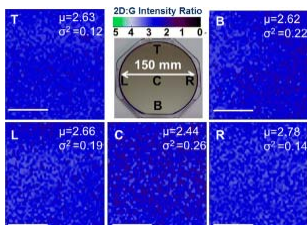
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## Carrier Mobility in Graphene

Abundance:  
 #1 O  
 #2 Si  
 ...  
 #15 C  
 ...  
 #35 Ga  
 ...  
 #53 Ge  
 #55 As  
 ...  
 #61 Sb  
 #67 In

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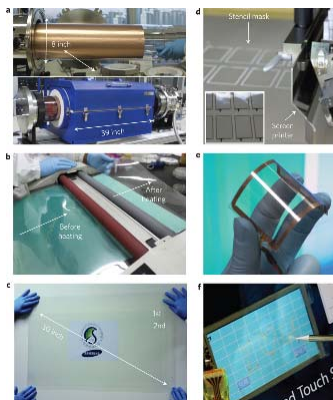
## Wafer Scale and Beyond



Rahimi, ACS Nano, 2014

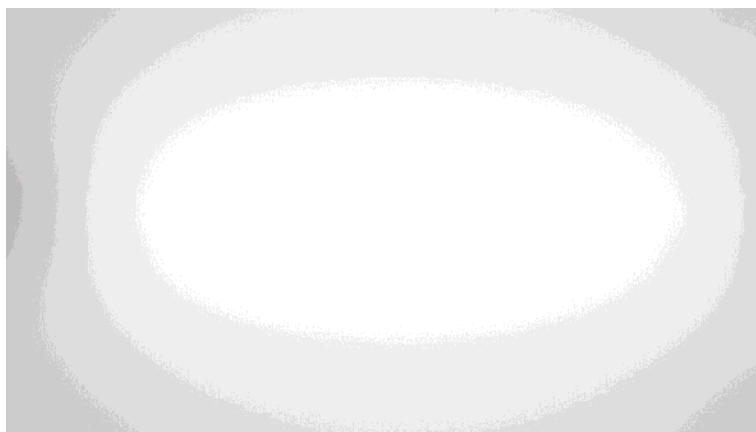
Average mobilities of  $\sim 2500$  cm<sup>2</sup>/V-s with 75% yield.

Roll to roll production on copper foils.



Bae, Nat. Nano. (2010)

## Water Purification





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## Biomedical

**nature**  
DNA SEQUENCING  
Enter the graphene nanopore

BIOMEDICAL NOISE  
Taking the good with the bad

THE NEW ENLIGHTENMENT  
Automated rules

PROSTATE  
ONCOLOGY  
The handbook of Luis Hayes

Graphene  
SiN<sub>4</sub>  
Si

Current (pA)  
Voltage (mV)

Membrane  
Membrane + nanopore

Event blockage (pA)  
Event duration (μs)

Constant e.c.d.  
200 μs

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**B**  
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## Biomedical

a) b) c) d) e)

100 μm 100 μm 100 μm 100 μm 100 μm

Lorenzoni, Sci. Reports (2013)

**B** With IGF-1

Day 2 Day 4

S G S G S G

Bajaj, Adv. Health. Materials (2013)

Planar graphene has been used to pattern and differentiate stem cells.  
Cell growth aligns with graphene patterns.  
Graphene enhances differentiation over supporting substrate.  
Mechanical crosstalk only investigated mechanism.

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## Composites

Large scale exfoliation and reduction of graphene oxide can be used to tune the electrical properties of graphene polystyrene composites.  
Stankovich, Nature (2006)

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## Composites

**(a)**

**(b)**

**(a)**

Material	Tensile Strength (MPa)
Pristine Epoxy	~55
Nanocomposite (0.1% wt SWNT)	~60
Nanocomposite (0.1% wt MWNT)	~65
Nanocomposite (0.1% wt GPL)	~75

**(b)**

Material	Experimental Data (GPa)	Theoretical Results (GPa)
Pristine Epoxy	~2.8	~2.8
0.1% wt MWNT	~2.8	~2.8
0.1% wt SWNT	~2.8	~3.2
0.1% wt GPL	~3.8	~3.2

Rafiee, ACS Nano, 2009

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## Energy Storage

Graphene-based materials for electrochemical energy-storage devices

Raccichini, Nature Materials (2015)

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## Energy Storage

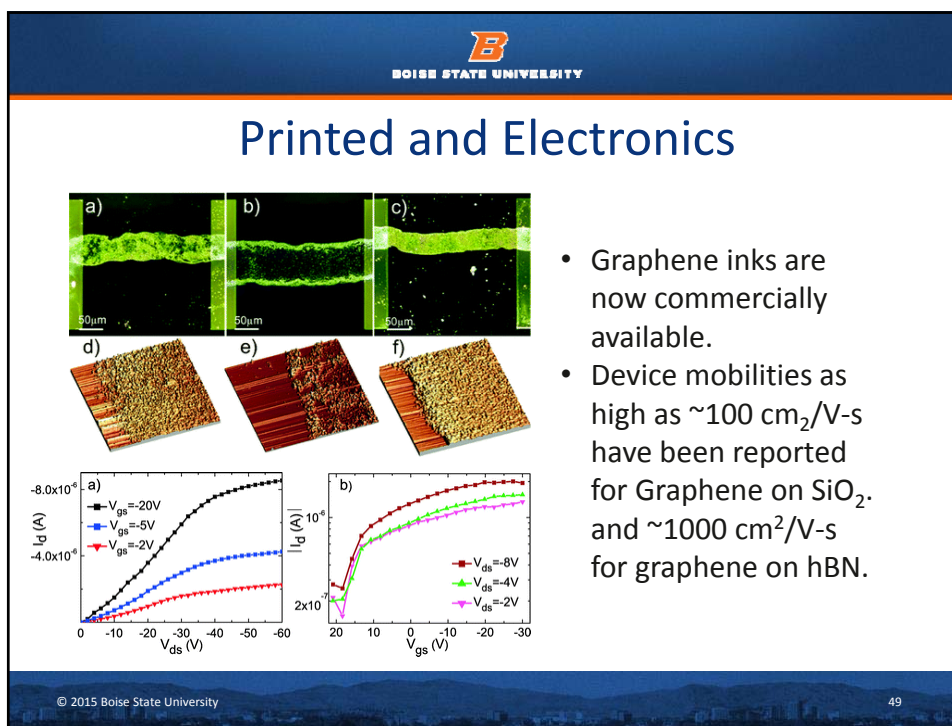
Randomly stacked layers of graphene nanoplatelets enhance charge storage in Li<sup>+</sup>, and Na<sup>+</sup> batteries. Enable flexible batteries.

Yakobson, J. Phys. Chem. Lett. (2013)

Graphene supercapacitors store as much energy as Ni metal hydride batteries (~70 Wh/kg)

Liu, Nano Lett. 2010

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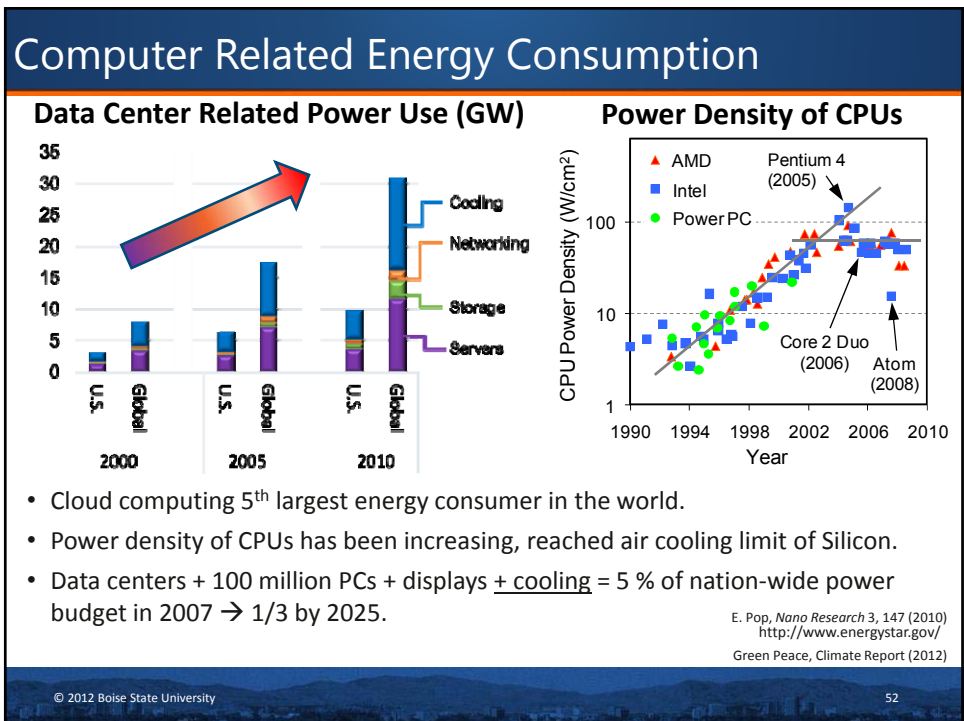
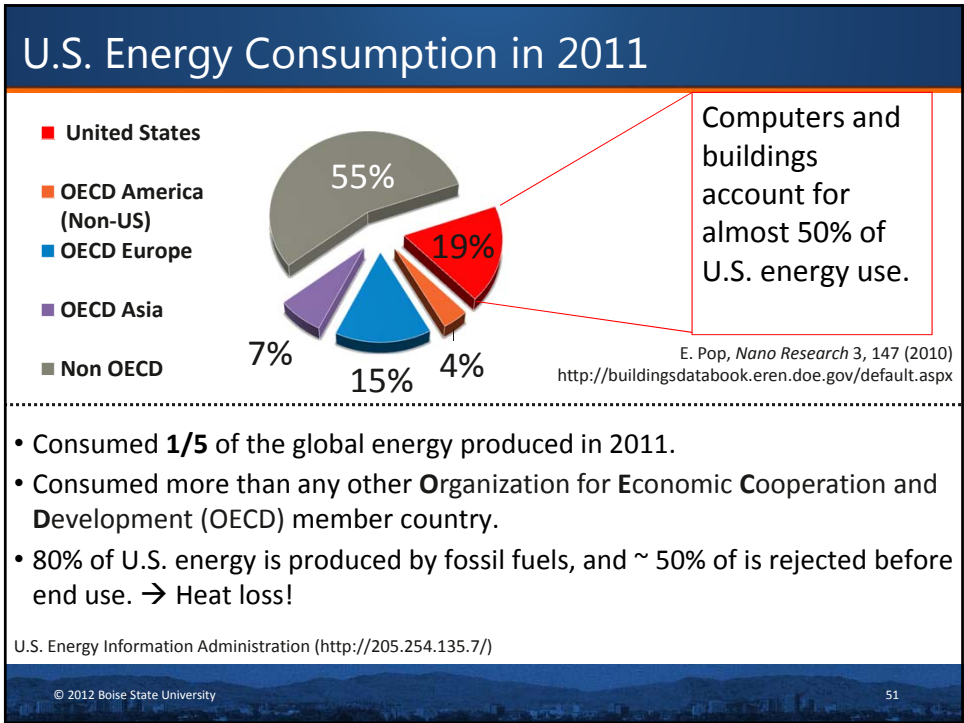


**B**  
BOISE STATE UNIVERSITY

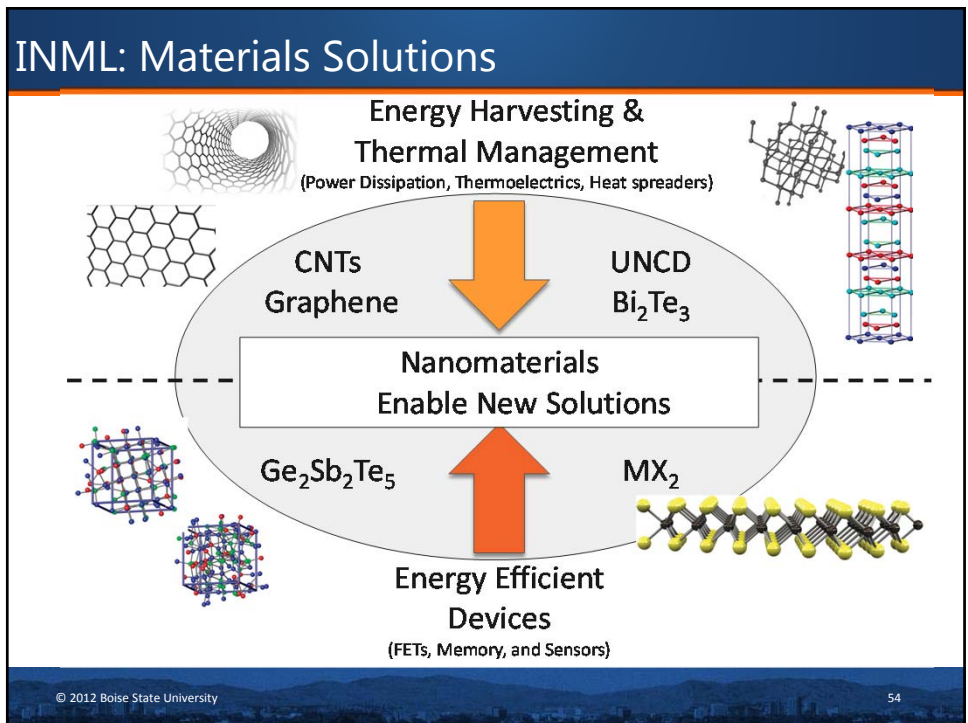
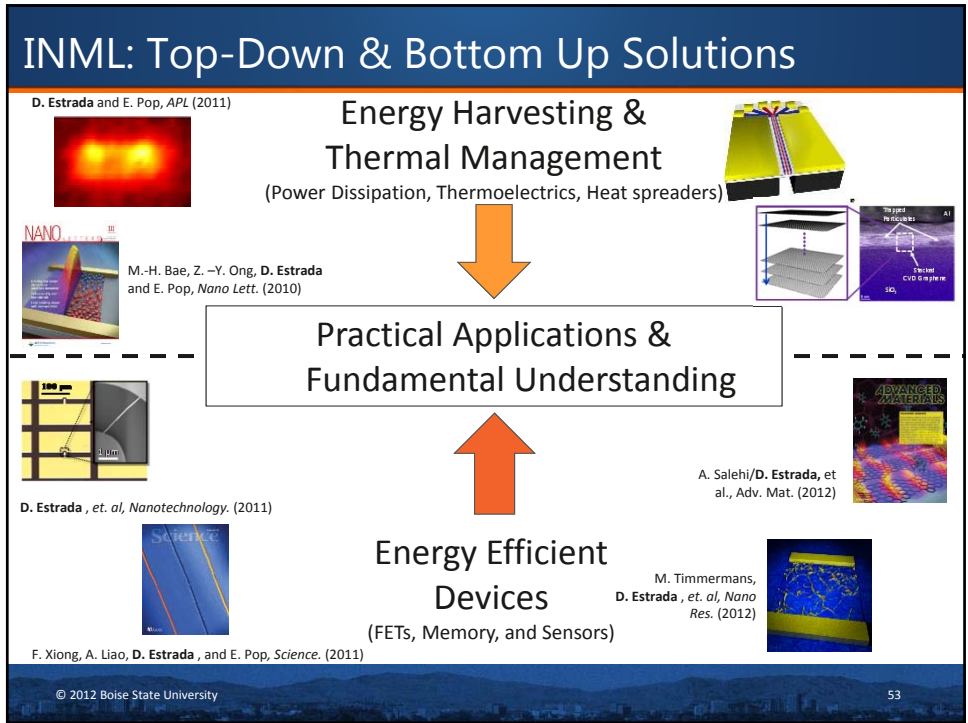
## Part 2: Atomic Layer Research in the Integrated NanoMaterials Lab

The figure shows three panels: a false-color scanning tunneling microscopy (STM) image of a graphene lattice with a color scale from 70 to 190 mV, a 3D visualization of a graphene lattice, and a scanning electron microscopy (SEM) image of a graphene device with a 500 μm scale bar.

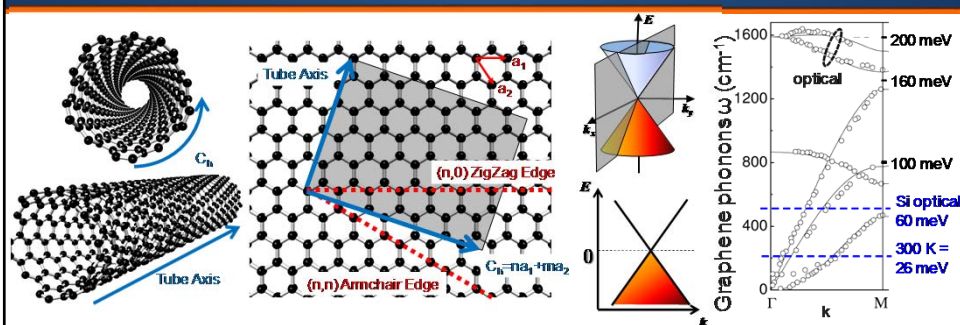
David Estrada  
Assistant Professor  
Materials Science and Engineering Department







## Properties of Carbon Nanomaterials



- Great electrical and thermal properties → low power dissipation!
  - Symmetric energy band, equal electron & hole mobility, 10-100x higher than Si: less scattering
  - Large optical phonon energy (0.18 eV vs. 0.06 eV in Si) : less scattering
  - Strong  $\sigma$  bonds → high thermal conductivity  $k \approx 20 \times k_{Si}$
- Gate tuning of carrier density and power dissipation

$$I \propto \eta \mu E$$

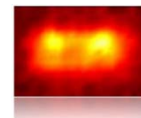
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55

## Outline of Part 2

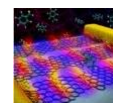
### Carbon Nanotube Thin-Film Transistors

- Direct Imaging of Power Dissipation
- Effects of CNT-CNT Junctions on Reliability



### Graphene Chemical Sensitivity

- Graphene Growth by Chemical Vapor Deposition
- Role of Linear Defects in Sensing



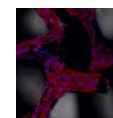
### Thermal Transport in CVD Graphene

- Suspended Thermometry Platform
- In-plane and cross-plane thermal conductivity measurements



### Nanobiotechnology Research in the INML

- Nanopores and graphene bioscaffolds

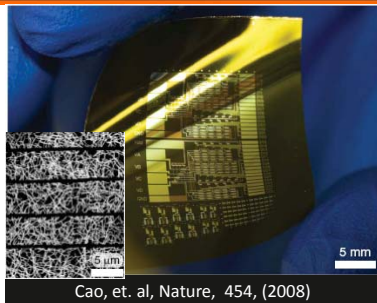


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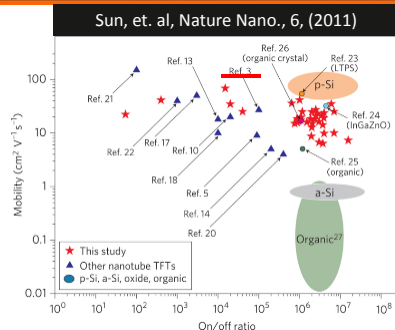
56



## Carbon Nanotube Network Transistors



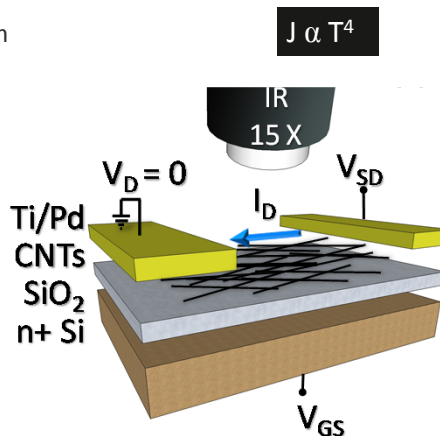
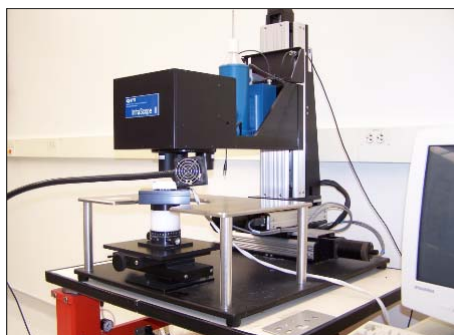
Cao, et. al, Nature, 454, (2008)



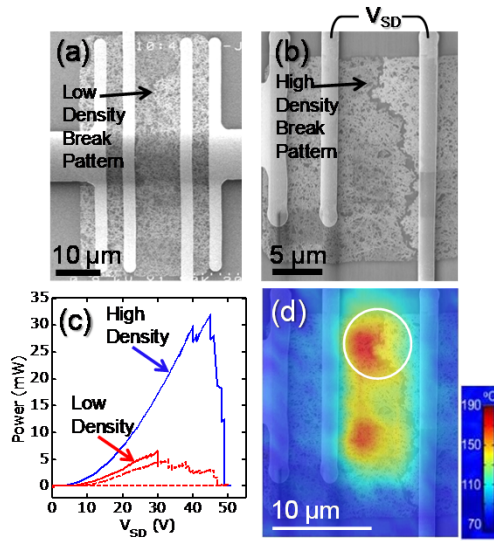
- CNT networks (CNNs) have lower performance than single CNTs ( $\mu$ ,  $I_{ON}/I_{OFF}$ ).
- CNN thin-film-transistors (TFTs) show promise as energy efficient display drivers.
- Low thermal conductivity substrates affect TFT performance  
→ Self heating and reliability.
- IR microscopy and electrical breakdown can be used to investigate power dissipation and reliability in CNN TFTs

## IR Imaging of CNN - TFTs

- QFI InfraScope Micro-Thermal Imager
- Specifications: 15X objective, spatial resolution 2.8  $\mu\text{m}$  and 0.1 K temperature resolution
- IR radiation of wavelength from 2 to 4  $\mu\text{m}$
- Stage temperature 70  $^{\circ}\text{C}$

D. Estrada and E. Pop, *Applied Physics Letters* 98, 073102 (2011)

## IR Imaging of CNN - TFTs



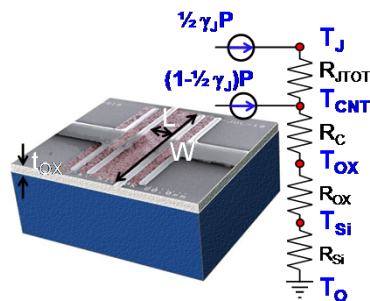
- Irregular breakdown pattern bears the imprint of the hotspot
- Channel temperature near breakdown << 600 °C
- Large drop in the P-V curve indicates break in the film
- Temperature profile signature of percolation

$$T_{BD} - T_0 = P_{BD} \times R_{TH}$$

D. Estrada and E. Pop, *Applied Physics Letters* **98**, 073102 (2011)

## Thermal Modeling of CNN-TFT

$$T_{BD} - T_0 = P_{BD} \times R_{TH}$$



$$g = 0.3 \text{ W K}^{-1} \text{ m}^{-1}$$

$$R_{TOT} = \frac{(T_{BD} - T_{CNT} - T_{OX} - T_{Si})}{\frac{1}{2} \gamma_J P_{BD}}$$

$$R_C = \frac{1}{g \times L_C}$$

$$R_{OX} = \frac{t_{OX}}{\kappa_{OX} A_C}$$

$$R_{Si} = \frac{1}{2\kappa_{Si} \sqrt{LW}}$$

600 °C
3.49 × 10 <sup>7</sup> K/W
104.5 °C
462.9 K/W
101.4 °C
4.46 × 10 <sup>3</sup> K/W
71.5 °C
223.6 K/W
70 °C

$$R_J \approx R_{TOT} \times n_J \times A \approx 4.4 \times 10^{11} \text{ K/W (2.27 pW/K), } 2.06 \times 10^{-6} \text{ m}^2 \text{ K W}^{-1}$$

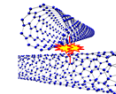
$$R_{TOT} \approx 3.38 - 3.50 \times 10^7 \text{ K W}^{-1} \text{ for } g = 0.05 - 0.6 \text{ W K}^{-1} \text{ m}^{-1}$$

Other work:  $3.3 \times 10^{11} \text{ K W}^{-1}$  (3 pW/K),  $1 \times 10^{-7} \text{ m}^2 \text{ K W}^{-1}$

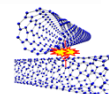
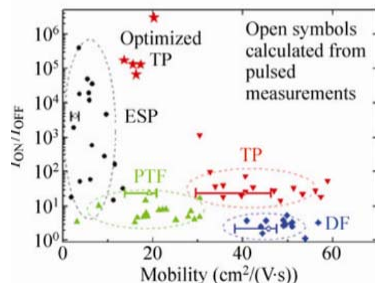
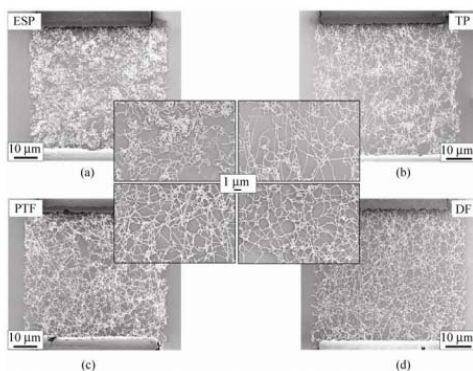
Zhong, et. al, PRB (2006)

Prasher, et al, PRL (2009)

D. Estrada and E. Pop, *Applied Physics Letters* **98**, 073102 (2011)



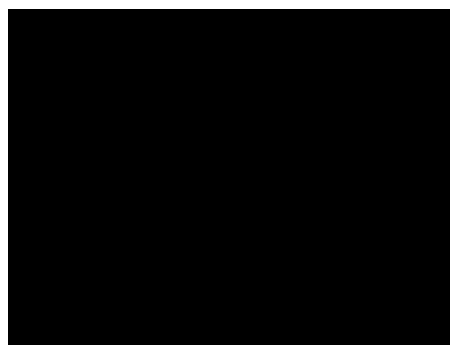
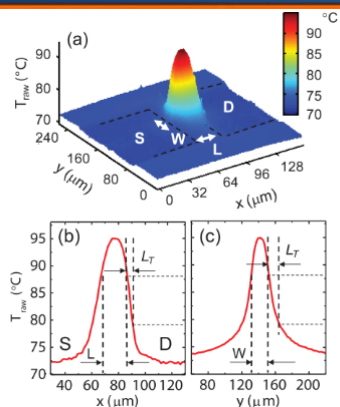
## CNT Network Morphology



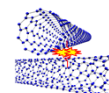
- Aerosol CVD of CNTs and deposition methods can result in drastically different network morphologies.
- Network morphology affects CNT-CNT junction areas → device mobility.
- Greater CNT alignment and reduced junction areas improve device performance.

M. Timmermans, D. Estrada, et. al, *Nano Research* 5, 307 (2012)

## Reducing Junction Resistance



- Use of monodisperse CNTs reduces the number Schottky barriers in the network → reduced resistance and uniform power dissipation.
- Thermal resistance of junctions can be used to solder together CNTs.

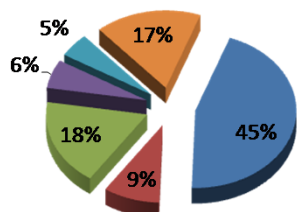


A. Behnam, et al., *ACS Nano* 7, 482 (2013)

J. W. Do, D. Estrada, et al., *Nano Letters* 13, 5844 (2013)

## Building Related Energy Consumption

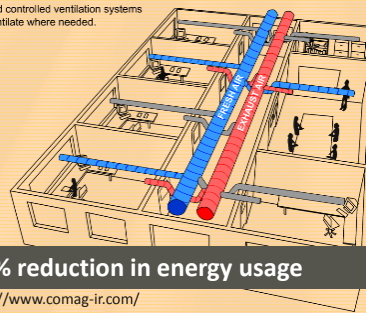
### Residential Energy Consumption By End Use



- Space Heating
- Space Cooling
- Water Heating
- Lighting
- Electronics
- Other

### Demand Controlled Ventilation (DCV)

Demand controlled ventilation systems only ventilate where needed.

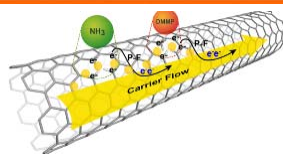


40% reduction in energy usage

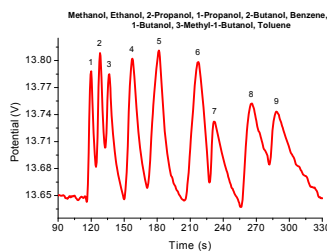
<http://www.comag-ir.com/>

- U.S. Buildings sector → 7% of global power consumption in 2010.
- 28 % of U.S. CO<sub>2</sub> emissions can be traced to HVAC systems.  
<http://buildingsdatabook.eren.doe.gov/>
- Adjust the ventilation rate based on pollutants level
- Potential to reduce heating & cooling costs
- *Refreshable, sensitive, and cheap, pollutant detectors needed*

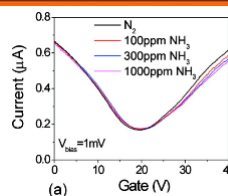
## Carbon Based Chemical Sensors



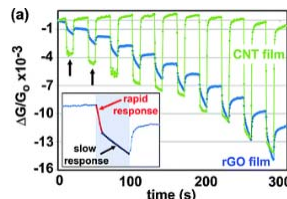
A. Salehi-Khojin, et al., *Science* (2010)



A. Salehi-Khojin, et al., *Nanoscale* (2011)



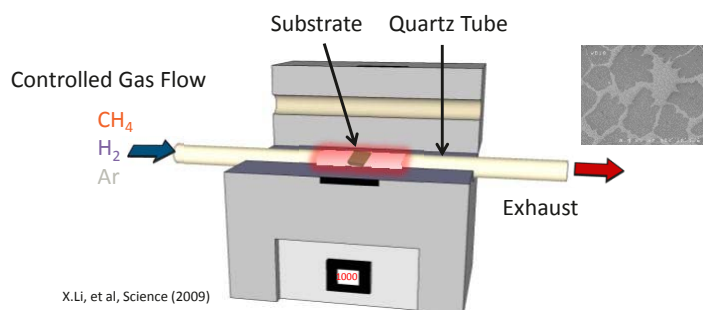
Y. Dan, et al., *Nano Letters* (2009)



J.T. Robinson, et al., *Nano Letters* (2008)

- 0-D defects in CNT sensors show to enhance sensitivity/selectivity  
→ Physics approach vs. chemical approach
- Graphene intrinsic response below that of CNTs, even for rGO films!

## Graphene Growth by CVD

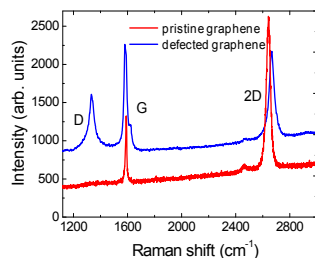
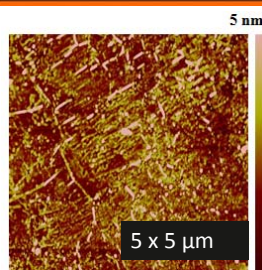


1" Quartz tube LPCVD furnace

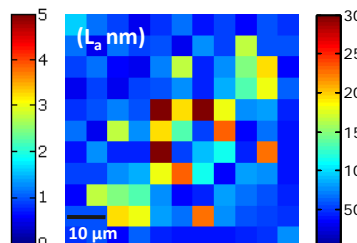
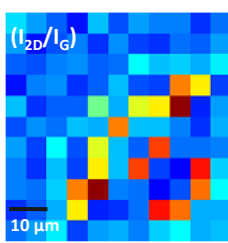
- Graphene grown directly on Cu foil at 1000 °C
- CVD growth of graphene → polycrystalline films
- Graphene is transferred by etching of Cu foil in  $\text{FeCl}_3$
- Films are annealed in  $\text{Ar}/\text{H}_2$  at 400 °C to remove residue

A.Salehi-Khojin / D. Estrada, et al., *Advanced Materials* **24**, 53-57 (2012)

## CVD Graphene is Inherently Defective



- AFM and Raman Spectroscopy reveal defective nature of CVD graphene
- Raman mapping shows our growth varies between mono and bilayer graphene
- The "crystallite" size is  $\sim 80$  nm
- Can line defects enhance chemical sensitivity?



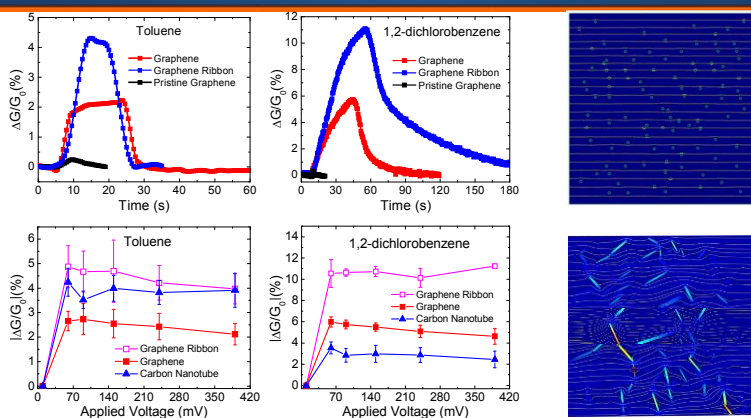
$$L_a \sim I_D/I_G$$

$$\# \text{Layers} \sim I_{2D}/I_G$$

A.Salehi-Khojin / D. Estrada, et al., *Advanced Materials* **24**, 53-57 (2012)



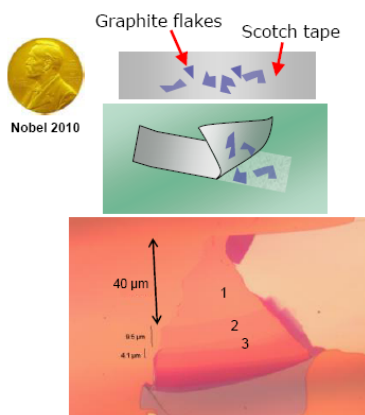
## Sensor Response to Trace Gas Vapors



- CVD graphene ribbons show high sensitivity to  $10^{14}$  molecules of toluene and  $10^{15}$  molecules of dichlorobenzene  $\rightarrow$  ppb sensitivity
- Confining current flow through linear defects allows polar molecules to modulate device conductance
- Defects in graphene enhance chemical sensitivity A.Salehi-Khojin / D. Estrada, et al., *Advanced Materials* 24, 53-57 (2012)

## Other Ways of Obtaining Graphene

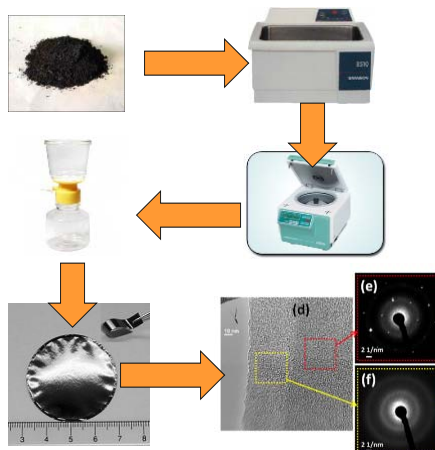
### Mechanical Exfoliation



Novoselov, K. S. et al., *Science* (2004)

Novoselov, X. et al., *Science* (2009)

### Chemical Assisted Exfoliation

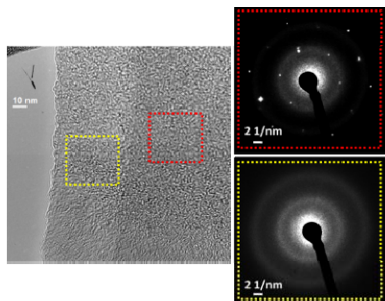


Y. Hernandez, et al., *Nat. Nano* (2008)

D. Li, et al., *Nat. Nano* (2008)



## Other Graphene Chemical Sensor Studies



APPLIED PHYSICS LETTERS 100, 033111 (2012)

### Chemical sensors based on randomly stacked graphene flakes

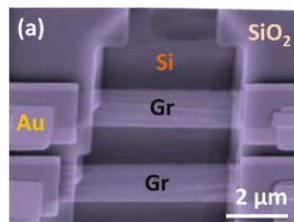
Amin Salehi-Khojin,<sup>1,7,a)</sup> David Estrada,<sup>2,3</sup> Kevin Y. Lin,<sup>1</sup> Ke Ran,<sup>4,5</sup> Richard T. Haasch,<sup>5</sup> Jian-Min Zuo,<sup>4,5</sup> Eric Pop,<sup>2,3,6</sup> and Richard I. Masel<sup>7,a)</sup>  
<sup>1</sup>Department of Chemical and Biomolecular Engineering, University of Illinois at Urbana-Champaign, Urbana, Illinois 61801, USA  
<sup>2</sup>Department of Electrical and Computer Engineering, University of Illinois at Urbana-Champaign, Urbana, Illinois 61801, USA  
<sup>3</sup>Micro and Nanotechnology Lab, University of Illinois at Urbana-Champaign, Urbana, Illinois 61801, USA  
<sup>4</sup>Department of Material Science and Engineering, University of Illinois at Urbana-Champaign, Urbana, Illinois 61801, USA  
<sup>5</sup>Materials Research Laboratory, University of Illinois at Urbana-Champaign, Urbana, Illinois 61801, USA  
<sup>6</sup>Beckman Institute, University of Illinois at Urbana-Champaign, Urbana, Illinois 61801, USA  
<sup>7</sup>Dioxide Materials, 60 Hazelwood Dr., Champaign, Illinois 61820, USA

NANO LETTERS

Letter  
pubs.aip.org/nanolett

### The Role of External Defects in Chemical Sensing of Graphene Field-Effect Transistors

B. Kumar,<sup>1</sup> K. Min,<sup>2</sup> M. Bashirzadeh,<sup>3</sup> A. Barati Farimani,<sup>2</sup> M. H. Bae,<sup>1,4</sup> D. Estrada,<sup>5</sup> Y. D. Kim,<sup>6</sup> P. Yasaei,<sup>1</sup> Y. D. Park,<sup>2</sup> E. Pop,<sup>1</sup> N. R. Aluru,<sup>7</sup> and A. Salchi-Khojin<sup>8,2</sup>

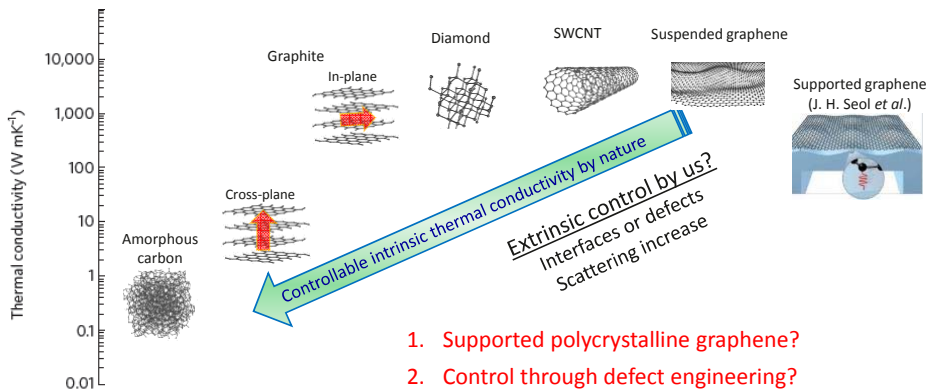


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## Thermal Conductivity of Carbon Allotropes

### Can we tune thermal transport properties ?



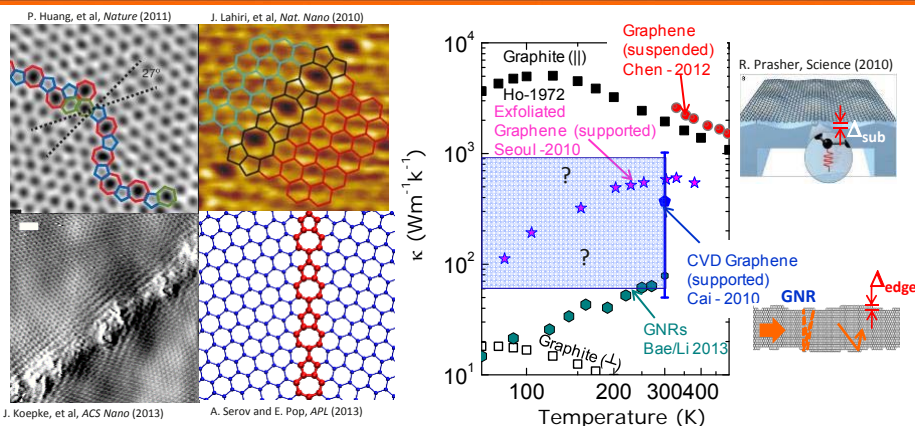
A. A. Balandin Nat. Mater. **10**, 569 (2011)  
 R. Prasher, Science **328**, 185 (2010)  
 J. H. Seol *et al.*, Science **328**, 213 (2010)  
<http://www.chemicool.com/elements/carbon.html>

1. Supported polycrystalline graphene?
2. Control through defect engineering?
3. Weaken cross-plane coupling?
4. Tune thermal anisotropy?
5. Transparent heat spreaders?

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## Linear Defects in Large Scale Graphene Films

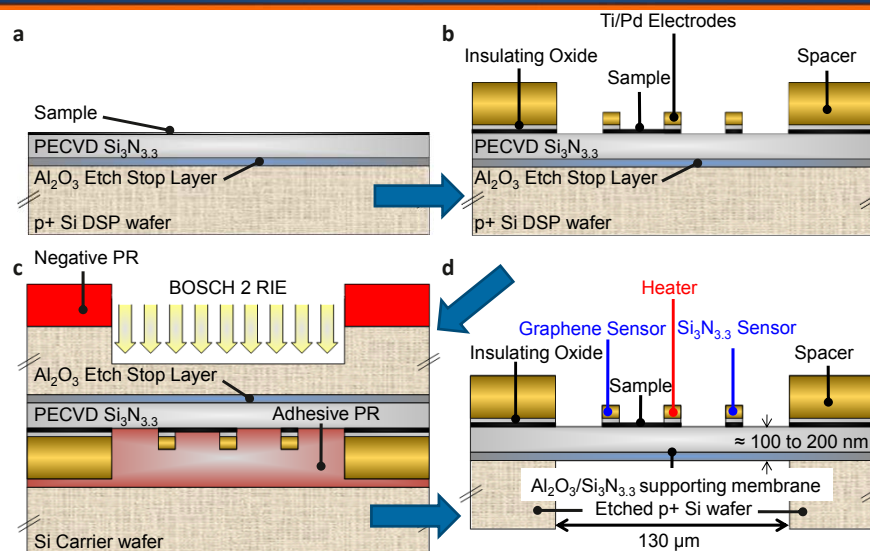


- Grain boundary structure varies greatly in polycrystalline CVD graphene films
- Significant carrier scattering observed by STM at grain boundaries
- Theory predicts significant phonon scattering at grain boundaries (depends on type)
- Extrinsic influences → tunable thermal transport in low-dimensional crystal

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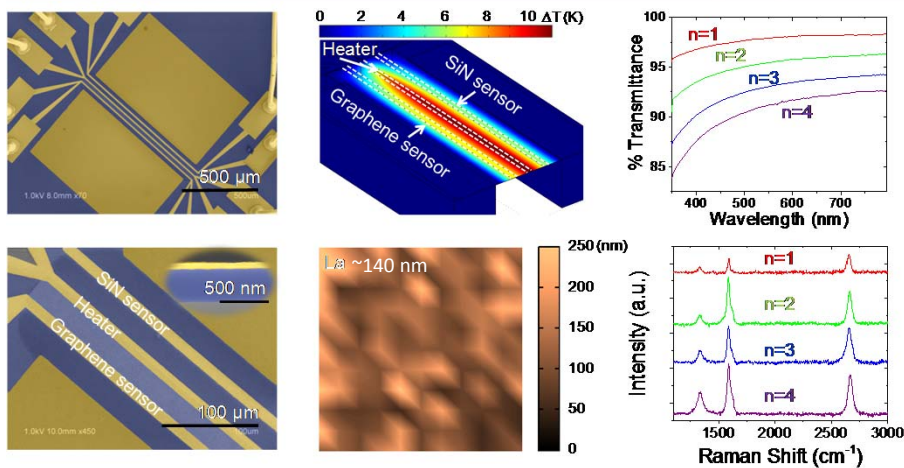
## Suspended Thermometry Platform Fabrication



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# Suspended Electrical Thermometry Platform

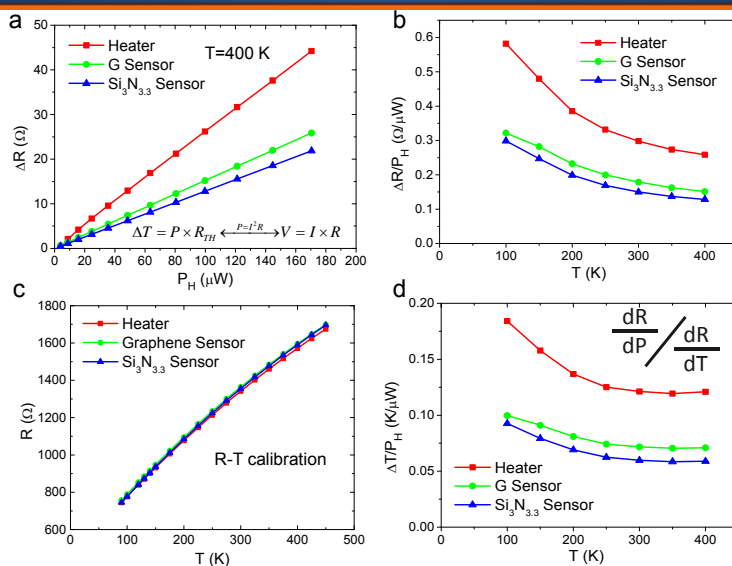


- Devices are wirebonded to a leaded chip carrier and placed inside a Janis VTSP for measurements (80 to 450 K)
- Graphene films are characterized by Raman, absorbance, AFM, and XPS.

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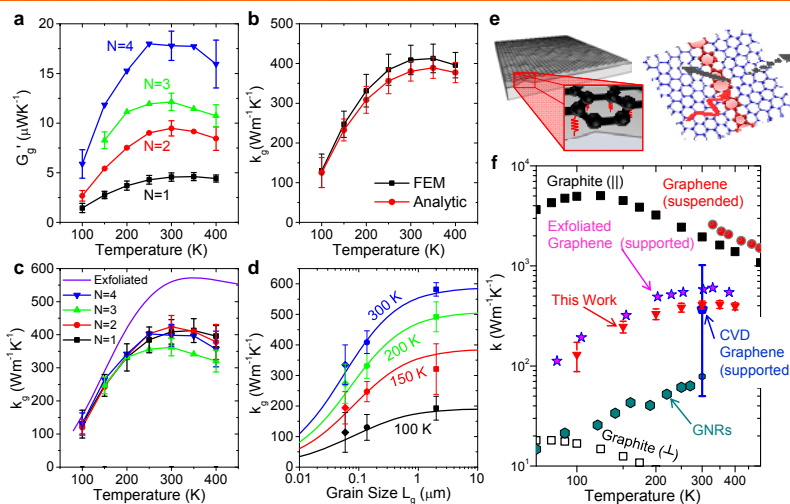
# Suspended Electrical Thermometry Platform



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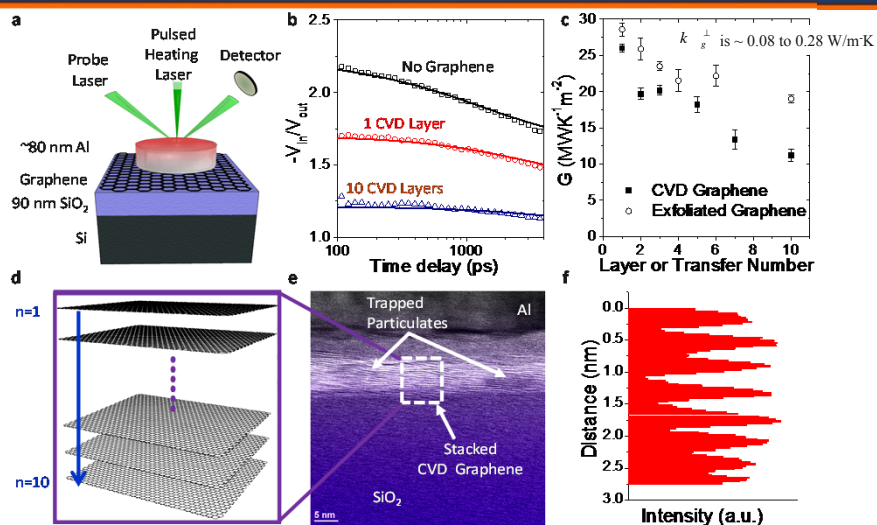
74

## Suspended Electrical Thermometry Platform



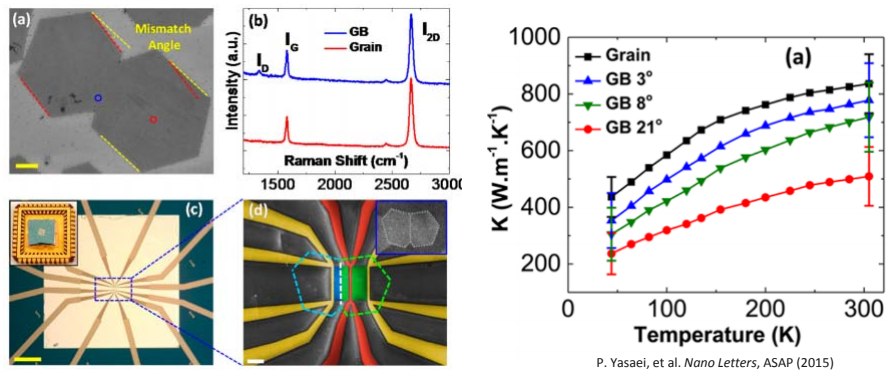
Thermal conductivity is limited by grain size and substrate scattering ( $N < 4$ )

## Cross-Plane Thermal Measurements



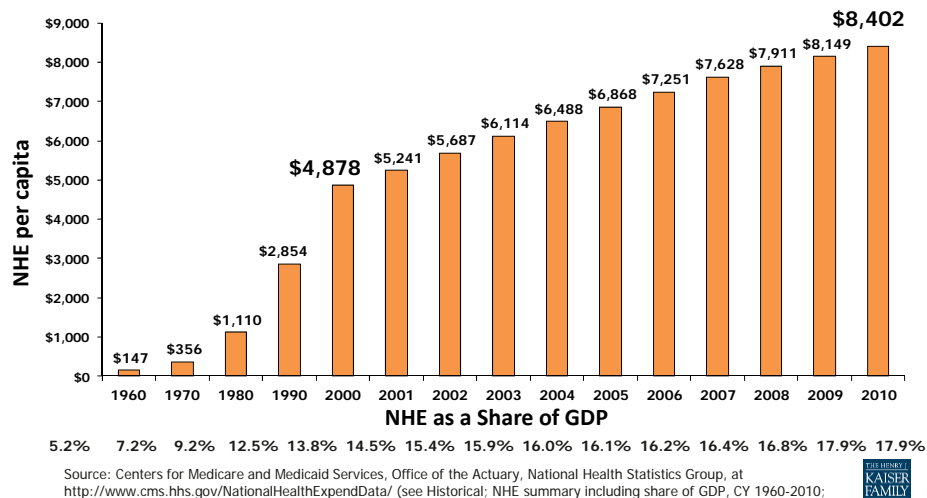
Cross-plane thermal conductivity is  $2 \times$  exfoliated  $\rightarrow$  tunable anisotropy

## Thermal Transport Across Grain Boundaries



- Direct electrical thermometry of CVD graphene with and without a GB reveals strong phonon scattering at GBs.
- Thermal conductivity decreases with increased GB misalignment → increased structural disorder near the GB.

## U.S. Healthcare Expenditures

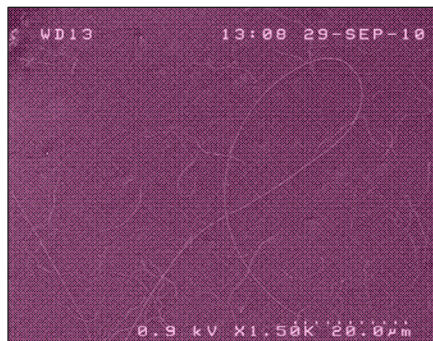


Projected to grow to \$13,708 in 2020 (~20% of GDP)



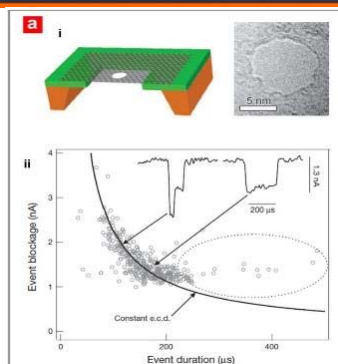
## It's not always the economy.....

- 7.6 million deaths (around 13% of all deaths) in 2008.
- About 70% of all cancer deaths occurred in low- and middle-income countries
- Deaths projected to continue to rise to over 13.1 million in 2030.
- High chance of cure if detected early and treated adequately

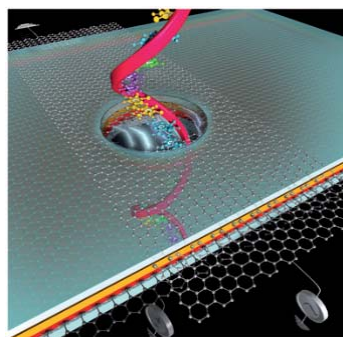


<http://www.who.int/>

## Nanopores in Graphene



Garaj et al., *Nature*, 2010

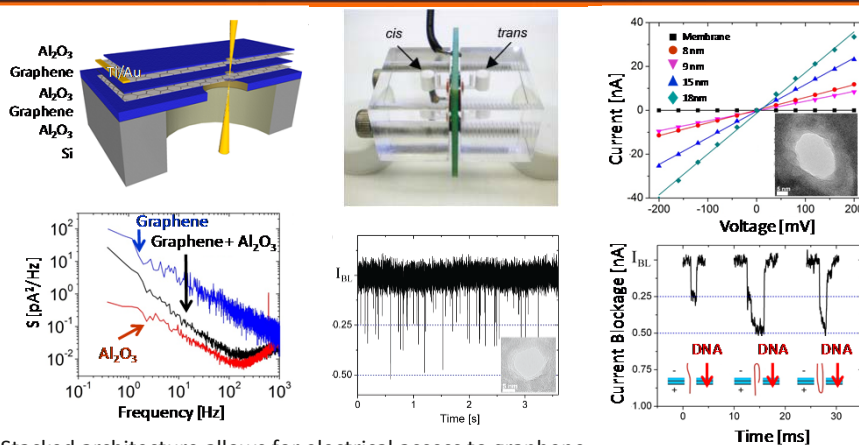


Venkatesan and Bashir, *Nat. Nano* 2011

- Interlayer spacing of graphite (graphene thickness)  $\approx 0.35$  nm
- Inter-nucleotide separation in ssDNA  $\approx 0.35 - 0.55$  nm
- Both were studied by Rosalind Franklin  $\sim 60$  years ago!
- Electrical access and new 2D materials enables new sensing modalities



## Stacked Dielectric-Graphene Architecture



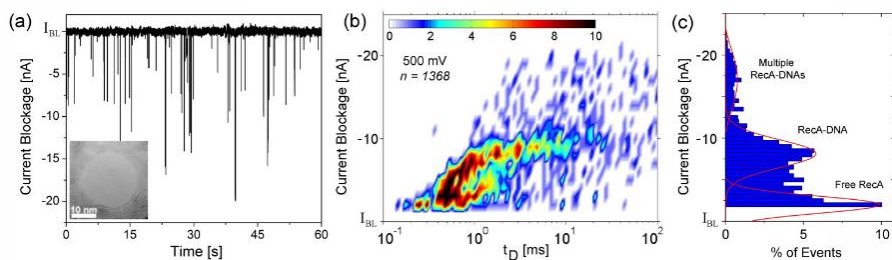
- Stacked architecture allows for electrical access to graphene
- Pore conductance varies with diameter  $\rightarrow$  lower  $1/f$  than graphene only (1 M KCl, 10 mM Tris, 1 mM EDTA, pH 8)
- $\lambda$ -DNA translocation and structure detected through ionic current blockades

B.M. Venkatesan, D. Estrada, et al., *ACS Nano* 6, 441 (2012)

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## DNA-RecA Protein Translocation



- Signature current blockades observed during DNA-protein translocation (25 nm pore, 400 mV, 1 M KCl, 10 mM Tris, 1 mM EDTA, pH 8)
- Unique current blockades correspond Free RecA, RecA DNA, and multiple RecA-DNA translocation events
- Proof of principle experiment shows possibility of using MBD proteins for early cancer detection

B.M. Venkatesan, D. Estrada, et al., *ACS Nano* 6, 441 (2012)

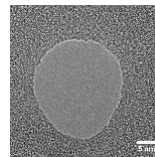
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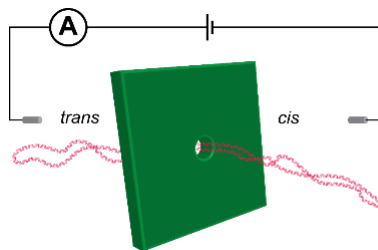
## EXPERIMENTAL CONDITIONS

### Nanopores:

- Drilled with JEOL 2010F TEM in 20 nm thick suspended  $\text{Si}_3\text{N}_4$  membrane TEM grid (Norcada)
- ~ 21 nm diameter nanopore sandwiched between two flow cells forming only electrical path between reservoirs



Nanopore  
TEM image

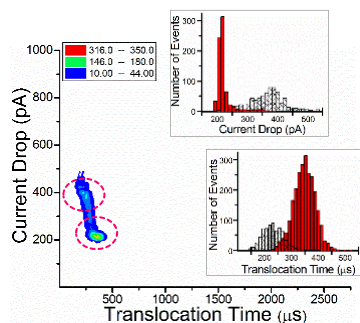


### Translocation:

- Reservoir electrolyte: 1M KCl, 1X TE buffer
- 7.5 kbp circular DNA: pTYB21 (NEB)
- Axopatch 200B amplifier, Digidata 1440 data acquisition system (Molecular Devices)
- 100 mV bias
- 10 kHz low pass hardware filter
- Added 50  $\mu\text{M}$  Ethidium Bromide (EtBr) solution to DNA solution to alter topology

## CIRCULAR DNA TRANSLOCATION EVENTS

Translocation of 7.5 kbp circular DNA suggests two distinct molecular conformations:



- Data analyzed using custom MATLAB routines and Origin 8.5
- Analyzed ~ 8000 events to determine the maximum current blockade and translocation time
- We can identify 2 distinct regions in the density plot and histograms which suggests 2 molecular conformations

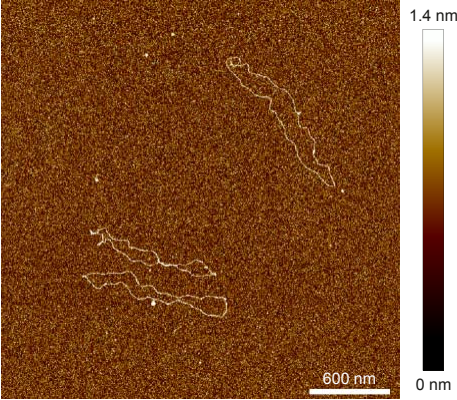
Region 2: supercoiled branched

Region 1: circular covalently closed

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## AFM IMAGES REVEAL DNA TOPOLOGY

- Optimized sample preparation
  - DNA incubated with NiCl<sub>2</sub> on fresh cleaved mica for 5 minutes
  - Rinsed samples with ultrapure H<sub>2</sub>O and gently dried with N<sub>2</sub>
- AFM imaging
  - Tapping mode, Bruker MultiMode 8 AFM
  - Image processing with NanoScope Analysis software
- 2D projection of the DNA in solution
- Representative AFM image demonstrate initial circular DNA topology

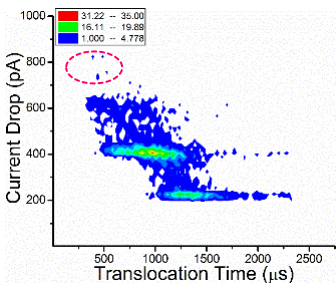



*Some branching, but generally uniform topology*

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## INTERCALATED CIRCULAR DNA TOPOLOGY: *REGION 4*





- Small population of events (~ 5%) with higher current drop: > 700 pA
- Multiple identifiable current drops

↓

*Further unwinding causing:*

- *additional branched structures*
- *secondary branching*



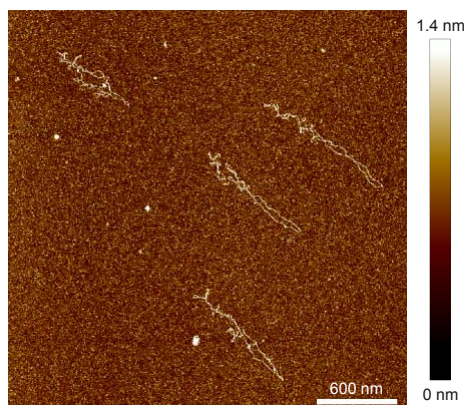
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## AFM IMAGES OF INTERCALATED DNA

### Circular DNA treated with EtBr exhibits significant branching

- Optimized sample preparation
  - DNA incubated with  $\text{NiCl}_2$  and EtBr on fresh cleaved mica for 5 minutes
  - Rinsed samples with ultrapure  $\text{H}_2\text{O}$  and gently dried with  $\text{N}_2$
- AFM imaging
  - Tapping mode, Bruker MultiMode 8 AFM
  - Image processing with NanoScope Analysis software
- Representative AFM image of intercalated circular DNA
- Under identical imaging conditions, observable structural changes to topology



## MOLECULAR DYNAMICS SIMULATIONS

### All-atom Molecular Dynamics simulations identify unwinding due to intercalation:

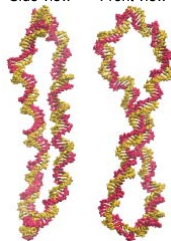


#### Simulation Conditions:

- 750,000 atoms
- 180 base-pair negatively supercoiled circular DNA (Lk = 14)
- 1M ion concentration

Initial circular DNA structure

Side view Front view



After treatment with 12 ethidium molecules for 100 ns

Side view Front view



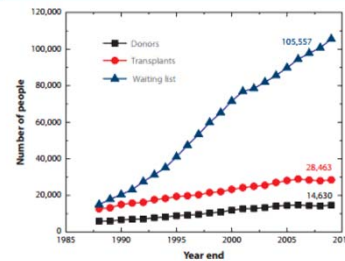
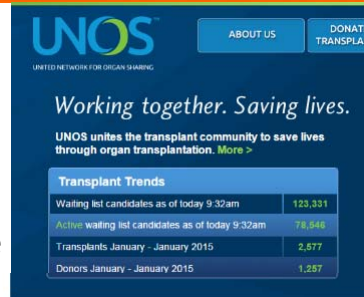
Ethidium initiates a positive supercoiling and unwinding of the circular DNA

Average total twist changes by  $160^\circ$



## The Need for Tissue Engineering

- Current waiting list for organ transplantation exceeds 123,000 (>78,000 active).
- Widening gap between patients and donors.
- Transplantation of vital organs is the only treatment for end stage organ failure.
- No such treatments exist for Traumatic Brain Injury (TBI) or neuromuscular disorders.
- New materials are needed to mimic cell-cell and cell-ECM interactions.

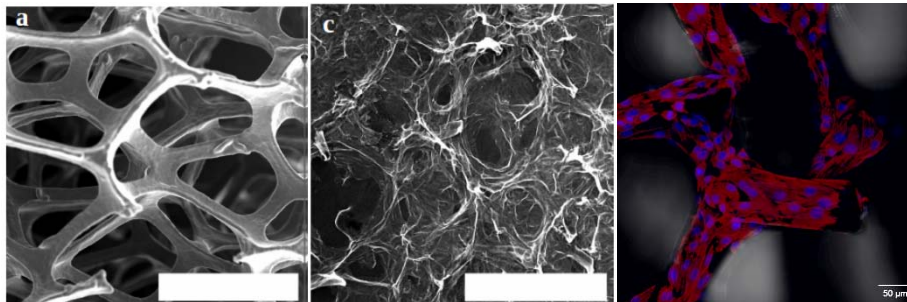


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P. Bajaj, Review of Biomedical Engineering (2014)

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## C2C12 Growth on Graphene Foam



- Growth of 2D materials can be extended to 3-dimensions on 3D transition metal foams.
- Selective etching removes metal scaffold → biocompatible 3D scaffold for cell growth
- C2C12 myoblasts can be seeded on laminin coated graphene foam.
- Confocal fluorescence microscopy of C2C12 cells shows high adhesion to foam walls:
  - Red - Alexa Fluor 546 phalloidin (F-actin)
  - Blue – DAPI
- Electrical stimulation and defects could play a role in cell growth and differentiation

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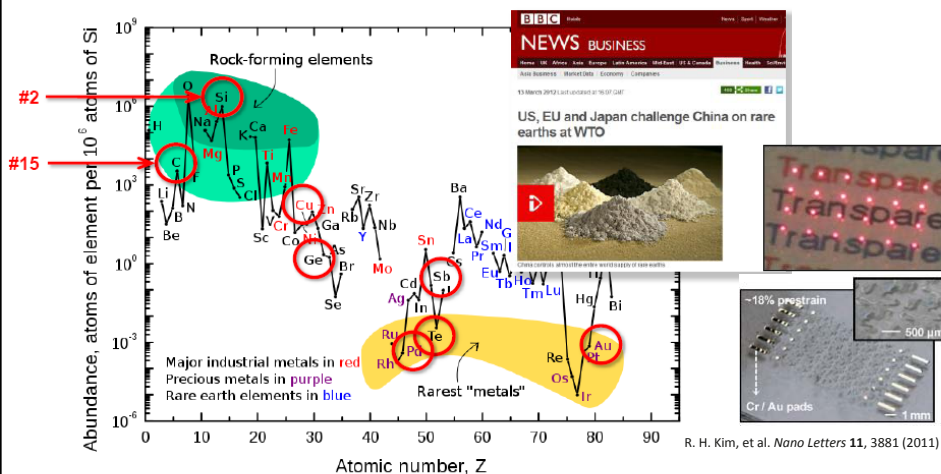
## Cell Growth for Neural Network

- Culture electrically responsive cells for developing functional engineered neural networks
- PC-12 rat pheochromocytoma cells
- Commonly used to model neurons
- Differentiate into neuron-like cells



Phase contrast image of PC-12 cells in laminin-coated dish. In media with 100ng/ml nerve growth factor.

## Using Carbon makes \$en\$



source: <http://pubs.usgs.gov/fs/2002/fs087-02>

Cost of Indium  $\text{kg}^{-1}$  has increased from ~\$94.00 to ~\$850 this decade.

(<http://www.metallbulletin.com/>)

  
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## THANK YOU

<http://coen.boisestate.edu/inml/>

Doctoral Committee:



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